Synthesis of Enantiopure Arylkainoids: Preparation of $(2S)\text{-}\bar{\Delta}^3$ -4-Phenylkainic Acid

Patrice Gill and William D. Lubell*

Département de chimie, Université de Montréal, C. P. 6128, Succ. **A,** *Montreal, Quebec, Canada H3C 3J7*

Received February 24, *1995*

The kainoid group of pyrrolidinedicarboxylic acids are attractive synthetic targets because of their structural complexity and because they exhibit potent and specific neuroexcitatory activity at glutamate receptors.^{1,2} Growing interest in glutamate pharmacology currently demands improved methodology to synthesize these receptor ligands for use as tools to study glutamate-mediated neurochemistry in order to develop memory-enhancing and neuroprotective therapeutics.³ In addition, the powerful excitatory action at insect neuromuscular junctions that is exhibited by kainoid derivatives has attracted interest in their synthesis for use as potential pesticides in agricultural chemistry. 4

The neuroexcitatory activity of the kainoids is contingent on the combination of their glutamate-like structure, 4-position substituent, and ring-substituent stereochemistries.³ For example, *trans-2-carboxy-3-pyrrolidineacetic* acid **2** (Figure 1) possesses the glutamate core structure characteristic of α -kainic acid and exhibits agonist activity at kainate-type and N-methyl-D-aspartate-type receptors.⁵ $(-)$ -Domoic acid, the causative agent of toxicity in cultured blue mussels responsible for outbreaks of food poisoning in Canada, possesses a conjugated diene at the 4-position and exhibits potent excitatory activity.^{1a,4,6} Modification of the aromatic 4-position substituents of the acromelic acids has led to 4-arylkainoids having greater excitatory activity than α -kainic acid.² On the other hand, inversion of the 4-position stereochemistry manifests reduced neuroexcitatory activity in allokainic acid.3

Recent reports of the use of 4-hydroxyproline to prepare arylkainoids prompted us to reveal our approach to synthesize enantiopure kainoid derivatives.^{2a,d} Our route provides 4-arylkainoids via regioselective enolization of a 4-oxoproline, C-alkylation, 0-triflylation, and Pd(0)-catalyzed cross-coupling to an arylboronic acid. We have synthesized (2s)-A3-4-phenylkainic acid **(1)** in eight steps from **(2S,4R)-4-hydroxyproline** in 39% overall yield and >99% enantiomeric purity. Although the stereochemistry at the 3- and 4-positions influences both the potency and the receptor selectivity of kainoid analogues,³ to our knowledge, we are the first to prepare a Δ^3 -kainoid

(3) (a) Reviewed in Hansen, J. J.; Krogsgaard-Larsen, P. *Med.* Res. *Rev.* 1990, 10, *55.* (b) Shinozaki, H. In *Kainic Acid as a Tool in Neurobiology;* McGeer, E. G., Olney, J. W., McGeer, P. L., Eds.; Raven

Figure 1. Representative kainoid analogues.

analogue. **By** restricting the conformations of both the 3- and 4-position substituents on a rigid vinylglycine structure, we have synthesized **1** as a novel probe for exploring glutamate receptors. In light of the efficient methodology to stereoselectively reduce double bonds,⁷ Δ^3 -analogues should be useful intermediates for the synthesis of isomers in order to study the relationships between kainoid stereochemistry and bioactivity.

Our strategy begins with hydroxyproline as a chiral educt that is submitted to a sequence of esterification,⁸ phenylfluorenation,⁹ and oxidation¹⁰ in order to furnish benzyl **4-oxo-N-(9-(9-phenylfluorenyl))prolinate (4)** in **77%** overall yield (Scheme **1,** 9-(9-phenylfluorenyl) = PhF1). Our experience with N -(PhFl)- α -amino ketones suggested that regioselective enolization and alkylation of **4** could be used to obtain 3-alkylprolines in high yields without racemization of the chiral α -center.¹¹ By providing a direct means to alkylate the 3-position, this improved approach eludes the formation of an oxoproline-derived enamine as well as the low to moderate yields from enamine alkylation.^{2d,12} We obtained a 2:1 diastereomeric mixture of benzyl **3-((methoxycarbonyl)methyl)-4** oxo-N-(PhF1)prolinates **6** in 91% yield after enolization of **4** at -78 °C with KN(SiMe₃)₂ (200 mol %) in a 0.06 M THF:DMPU (9:l) solution and treatment with methyl bromoacetate (220 mol **%).'3**

We examined three approaches in order to synthesize 4-arylprolines. Initially, benzyl 4-hydroxy-N-(PhFl)prolinate **(3)** was converted into tosylate **6** on treatment with TsCl and pyridine. Although arylcuprates add to 4-(to**syloxyl-N-(acy1)prolines** providing 4-arylprolines with retention of configuration, 2a,14 we could not obtain benzyl **4-phenyl-N-(PhFl)prolinate** using this chemistry on **6.** We then tried to add organometallic reagents to ketone **4.** Arylcerium reagents react with 4 -oxo- \bar{N} -(acyl)prolines in

(8) Zervas, L.; Winitz, M.; Greenstein, J. P. J. *Org. Chem.* 1957,22, **1515.**

(9) Lubell, W.; Rapoport, H. *J. Org. Chem.* 1989,54, 3824. **(10)** Mancuso, A. J.; Huang, S. L.; Swern, D. *J.* Org. *Chem.* 1978,

- 43, 2480.
- (11) Lubell, W. D.; Rapoport, H. J. Am. Chem. Soc. 1988, 110, 7447.
(12) Holladay, M. W.; Lin, C. W.; May, C. S.; Garvey, D. S.; Witte, D. G.; Miller, T. R.; Wolfram, C. A. W.; Nadzan, A. M. J. Med. Chem.
- 1991,34,455. (13) (a) We have employed a variety of electrophiles to alkylate 4 after observing specific incorporation of deuterium at the proline 3-position ($\delta = 2.3$ and 2.4 ppm) on treatment of the enolate with CD_3 -OD. R. Sharma and W. D. Lubell, manuscript in preparation. **(b)** In a related synthesis of dihydroxypyrrolidines, the sodium enolate of related synthesis of dihydroxypyrrolidines, the sodium enolate of methyl δ -oxo-N-(PhFl)prolinate has been oxidized to a 3-hydroxy derivative, see: Blanco, M. J.; Sardina, F. J. *Tetrahedron Lett.* 1994, 35, 8493.

(14) Thottathil, J. K.; Moniot, J. L. *Tetrahedron Lett.* 1988,27, 151.

^{(1) (}a) Kainoid syntheses are reviewed in Hashimoto, K.; Skirahama, H. *Trends Org. Chem.* 1991, 2, 1. (b) Oppolzer, W.; Robbiani, C.; Bättig,
K. *Tetrahedron* 1984, 40, 1391. (c) Cooper, J.; Knight, D. W.; Gallagher,
P. T. J. *Chem. Soc., Perkin Trans. 1*_1992, 553. (d) Hatakeyama, S.; Sugawara, **IC;** Takano, S. J. *Chem.* SOC., *Chem. Commun.* **1993,** 125.

Sugawara, K., Tanano, S. J. Crg. Chem. 30c, Chem. 1994, 59, 6968.

(e) Yoo, S.-e.; Lee, S. H. J. Org. Chem. 1994, 59, 6968.

(2) Recent syntheses of arylkainoids include the following. (a)

Hashimoto, K.; Shirahama, H. Te Rudolph, M. *Tetrahedron Lett.* 1994, 35, 6163.

Press: New York, 1978; pp 17–35.

(4) Maeda, M.; Kodama, T.; Saito, M.; Tanaka, T.; Yoshizumi, H.;
Nomoto, K; Fujita, T. *Pestic. Biochem. Physiol.* **1987**, 28, 85.

(5) Tsai, C.; Schneider, J. A.; Lehmann, J. *Neurosci. L*

^{298.}

⁽⁶⁾ Wright, J. L. C.; Boyd, R. K.; De Freitas, A. S. W.; Falk, M.; Foxall, R. A.; Jamieson, W. D.; Laycock, M. V.; McCulloch, A. W.; McInnes, A. G.; Odense, P.; Pathak, V. P.; Quilliam, M. A.; Ragan, M. A.; Sim, P. G.; Thi A.; Dewar, D. *Can. J. Chem.* 1989,67,481.

⁽⁷⁾ Recent reviews include the following. (a) Noyori, R. *Tetrahedron* 1994,50, 4259. (b) Brown, J. M. *Angew. Chem., Int. Ed. Engl.* 1987, *26,* 190.

Scheme 1. Alkylation and Nucleophilic Addition onto 4-Oxoproline 4

better yields, providing cleaner products than aryllithium
and arylmagnesium reagents.^{2d,15,16} In our hands, 4-oxo-N-(PhF1)proline **4** reacted diastereoselectively with PheCeCl₂ (150 mol %) in THF at -60 °C and $(2S, 4R)$ -4phenyl-4-hydroxyproline **7** was isolated in 60% yield. The *4R* stereochemistry was assigned initially on observation of the hydroxyl group 'H NMR signal which was downfield shifted due to hydrogen bonding to the ester carbonyl. Additional support for the *cis* relationship between the phenyl ring and α -proton was obtained from consideration that the a-proton shared vicinal and W coupling constants with the downfield shifted β - and δ -protons. The 3-alkyl substituent had a detrimental affect on the addition of PheCeClz to **5** and benzyl **3-((methoxycarbonyl)methyl)-4-phenyl-4-hydroxy-N-(Ph-**Fllprolinate was not isolated. Since Pd(0)-catalyzed cross-coupling of arylboronic acids and vinyl triflates furnishes styrenes possessing amino groups and tetrasubstituted olefins, 17 we explored this third alternative to synthesize 4-arylprolines.

Vinyl triflate *8* was prepared in good yield by enolization of 4-oxo-N-(PhFl)proline 4 with $KN(SiMe₃)₂$ in THF and O-acylation with $\mathrm{PhN(Tf)_2}$ (Scheme 2). 17a Palladiumcatalyzed cross-coupling of 8 with PhB(OH)₂ (110 mol %) was performed using 5 mol % of Pd(PPh₃)₄ in a two-phase system of DME and 2 N Na₂CO₃ with LiCl (300 mol %) at 70 "C for 1 h and gave styrene **9** in 72% yield.17' The vinyl proton signal in the spectrum of styrene **9** is shifted 0.3 ppm downfield from its position in the spectrum of triflate *8.* The stage was set to see if tetrasubstituted olefin **11** could be prepared from 3-((methoxycarbony1) **methyl)-4-oxo-N-(PhFl)proline** *(5).* Vinyl triflate **10** was prepared from **5** using the above conditions in 92% yield. At first, the cross-coupling did not furnish **11** and triflate **10** was recovered unchanged. Protected A3-phenylkainoid **11** was obtained in 92% yield by changing from DME to toluene and heating with $PhB(OH)₂ (300 mol %)$ at 95 "C for 6 h.17f Deprotection of **11** was accomplished in two steps. The esters were first hydrolyzed using 2 N sodium hydroxide (2500 mol %) in aqueous EtOH at **55** "C for 24 h to provide **(2S)-A3-4-phenyl-N-(PhFl)kainic** acid **(12)** in 78% yield after chromatography. Finally, solvolysis of the PhFl group using a 0.02 M solution of 1:1:40 TFA anisole: $\overline{\text{CH}}_2\text{Cl}_2$ at rt for 24 h provided 1 as a white solid in 82% yield after removal of the hydrocarbon impurities by trituration with hexane.18

The enantiomeric purity of **1** was ascertained after conversion to (R) - and (S) -N- $(\alpha$ -methylbenzyl)urea dimethyl esters 14. Esterification of 1 with MeOH^{HCl} and evaporation provided dimethyl ester hydrochloride **13** that was acylated quantitatively with either *(R)-* or *(8)-*

1989, 110, 193398g.

(16) Krapcho, J.; Turk, C.; Cushman, D. W.; Powell, J. R.; DeForrest, J. M.; Spitzmiller, E. R.; Karanewsky, D. S.; Duggan, M.; Rovnyak, G.; Schwartz, J.; Natarajan, S.; Godfrey, J. D.; Ryono, D. E.; **Miyaura, N.; Suzuki, A.** *J. Org. Chem.* **1993, 58, 2201.** *(0* **Huth, A.; Beetz, I.; Schumann, I.** *Tetrahedron* **1989,45, 6679.**

(18) Less than 1% trifluoroacetate was detected in the 19F **NMR spectrum** of **1.**

Scheme 2. Synthesis of Δ^3 **-4-Phenylkainic Acid (1) via Pd-Catalyzed Cross-Coupling on Vinyl Triflate 10**

 α -methylbenzyl isocyanate in THF with triethylamine.¹⁹ Observation of the diastereomeric methyl ester singlets by 300 MHz ¹H NMR spectroscopy in C_6D_6 during incremental additions of the opposite isomer demonstrated **14** to be of >99% de. Hence **1** is presumed to be of >99% enantiomeric purity.

We have developed an efficacious method to synthesize enantiopure arylkainoid derivatives. $(2S)$ - Δ ³-4-Phenylkainic acid **(1)** was synthesized in eight steps and 39% overall yield from **(25,4R)-4-hydroxyproline.** Rapid assembly of the trisubstituted pyrrolidine is achieved by regioselective enolization of 4-oxo-N-(PhFl)prolinate **4,** C-alkylation, 0-triflylation, and Pd(0)-catalyzed crosscoupling to an arylboronate. We are now examining diastereoselective olefin reduction as well as the bioactivity of the Δ^3 -4-phenylkainate derivatives. Because a variety of alkyl halides and arylboronic acids can be employed in this strategy, our method exhibits great potential for the synthesis of many kainoid analogues for the study of glutamate-mediated neurotransmission.

Acknowledgment. This research was supported in part by the NSERC of Canada and the Ministère de l'Éducation du Québec. W.D.L. thanks Bio-Méga/Boehringer Ingelheim Recherche Inc. for a Young Investigator Award.

Supplementary Material Available: Experimental details as well as lH and lSC NMR spectra of l, 6, and 7-12 and lH NMR spectra of 13 and 14 (23 pages).

509503705

⁽¹⁵⁾ (a) Thottathil; J. K. US. Pat. 4,734,508; *Chem. Abstr.* **1988, 109,55239j. (b) Thottathil; J.** K **Ger. Offen. DE 3,814,663;** *Chem. Abstr.*

⁽¹⁹⁾ Ibrahim, H. H.; Lubell, W. D. *J. Org. Chem.* **1993, 58, 6438.**